

## Evolution of the Cytochrome *c* Oxidase Proton Pump

Siegfried M. Musser, Sunney I. Chan\*

Division of Chemistry and Chemical Engineering, Arthur Amos Noyes Laboratory of Chemical Physics, California Institute of Technology, Pasadena, CA 91125, USA

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**Abstract.** The superfamily of quinol and cytochrome *c* terminal oxidase complexes is related by a homologous subunit containing six positionally conserved histidines that ligate a low-spin heme and a heme–copper dioxygen activating and reduction center. On the basis of the structural similarities of these enzymes, it has been postulated that all members of this superfamily catalyze proton translocation by similar mechanisms and that the Cu<sub>A</sub> center found in most cytochrome *c* oxidase complexes serves merely as an electron conduit shuttling electrons from ferrocycytochrome *c* into the hydrophobic core of the enzyme. The recent characterization of cytochrome *c* oxidase complexes and structurally similar cytochrome *c*:nitric oxide oxidoreductase complexes without Cu<sub>A</sub> centers has strengthened this view. However, recent experimental evidence has shown that there are two ubiquinone(ol) binding sites on the *Escherichia coli* cytochrome *bo*<sub>3</sub> complex in dynamic equilibrium with the ubiquinone(ol) pool, thereby strengthening the argument for a Q(H<sub>2</sub>)-loop mechanism of proton translocation [Musser SM et al. (1997) *Biochemistry* 36:894–902]. In addition, a number of reports suggest that a Q(H<sub>2</sub>)-loop or another alternate proton translocation mechanism distinct from the mitochondrial *aa*<sub>3</sub>-type proton pump functions in *Sulfolobus acidocaldarius* terminal oxidase complexes. The possibility that a primitive quinol oxidase complex evolved to yield two separate complexes, the cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes, is explored here. This idea is the basis for an evolutionary

tree constructed using the notion that respiratory complexity and efficiency progressively increased throughout the evolutionary process. The analysis suggests that oxygenic respiration is quite an old process and, in fact, predates nitrogenic respiration as well as reaction-center photosynthesis.

**Key words:** Cytochrome *c* oxidase — Ubiquinol oxidase — Nitric oxide reductase — Proton pump — Q(H<sub>2</sub>)-loop — Evolutionary tree — Respiratory efficiency — Respiratory complexity

### Introduction

The many similarities between the *Escherichia coli* cytochrome *bo*<sub>3</sub> complex and the *aa*<sub>3</sub>-type cytochrome *c* oxidase complexes led to the recognition of a superfamily of terminal oxidase complexes related in structure and function. All of the enzymes in this superfamily contain a low-spin, six-coordinate heme which mediates electron input to a heme–copper dioxygen activating and reduction center. This superfamily is divided into two families: the quinol (typically, ubiquinol) oxidase complexes and the cytochrome *c* oxidase complexes (Table 1). The variations within the families are mainly in the number and type of hemes present in the various enzymes. In general, all members of this superfamily of terminal oxidase complexes are thought to translocate protons against a transmembrane potential gradient (although this has not been proven in all cases) and all appear to catalyze dioxygen activation and reduction at a heme–copper binuclear center (García-Horsman et al. 1994a; Musser et al. 1995; van der Oost et al. 1994).

\* Present address: Division of Biological Sciences, Section of Plant Biology, University of California, Davis, Davis, CA 95616, USA  
Correspondence to: S.I. Chan; e-mail: Chans@cco.caltech.edu

**Table 1.** Representative members of the superfamily of ubiquinol and cytochrome *c* terminal oxidase complexes<sup>a</sup>

Type	Cu <sub>A</sub> present? <sup>b</sup>	Translocates protons?	Hemes <i>c</i> ?	Species	Reference(s)
Quinol oxidase complexes					
<i>bo</i> <sub>3</sub> <sup>c</sup>	No	Yes	No	<i>E. coli</i>	Chepuri et al. 1990; Minghetti et al. 1992; Puustinen et al. 1989, 1991
<i>ba</i> <sub>3</sub>	No	No	No	<i>A. acetii</i>	Fukaya et al. 1993; Matsushita et al. 1990
	Yes	Yes	No	<i>S. acidocaldarius</i> <sup>d</sup>	Lübben et al. 1994; Schäfer et al. 1996b
<i>bb</i> <sub>3</sub>	No	Yes	No	<i>P. denitrificans</i>	de Gier et al. 1996; Solioz et al. 1982; van der Oost et al. 1991a
	No	ND <sup>e</sup>	No	<i>B. japonicum</i>	Surpin et al. 1996
<i>aa</i> <sub>3</sub>	No	Yes	No	<i>B. subtilis</i>	Lauraeus et al. 1991; Santana et al. 1992
	No	Yes	No	<i>S. acidocaldarius</i> <sup>d</sup>	Anemüller et al. 1992; Anemüller and Schäfer 1989, Schäfer 1990; Gleißner et al. 1997; Lübben et al. 1992
Cytochrome <i>c</i> oxidase complexes					
<i>aa</i> <sub>3</sub> <sup>f</sup>	Yes	Yes	No		
<i>ba</i> <sub>3</sub>	Yes	ND	No	<i>T. thermophilus</i>	Zimmerman et al. 1988
<i>caa</i> <sub>3</sub>	Yes	Yes	1	<i>T. thermophilus</i>	Büse et al. 1989; Fee et al. 1988; Mather et al. 1993
	ND	Yes	1	<i>Bacillus stearothermophilus</i>	de Vrij et al. 1989; Sone and Fujiwara 1991
	Yes	Yes	1	Thermophilic <i>Bacillus</i> PS3 <sup>g</sup>	Ishizuka et al. 1990; Sone and Hinkle 1982; Sone and Yanagita 1984; Sone et al. 1988
	Yes	ND	1	<i>B. subtilis</i>	Lauraeus et al. 1991; Saraste et al. 1991; van der Oost et al. 1991b
<i>cbb</i> <sub>3</sub>	No	ND	3	<i>R. capsulatus</i>	Gray et al. 1994; Thöny-Meyer et al. 1994
	No	ND	3	<i>R. sphaeroides</i>	García-Horsman et al. 1994b
	No	Yes	3	<i>P. denitrificans</i>	de Gier et al. 1994, 1996
	No	ND	3	<i>B. japonicum</i>	Preisig et al. 1993; Zufferey et al. 1996

<sup>a</sup> This table includes the more well-characterized heme-copper oxidase complexes. See García-Horsman and co-workers (1994a) for a more expanded listing.

<sup>b</sup> The presence or absence of a Cu<sub>A</sub> site has not been definitively confirmed in all cases. A best guess is made based on the present literature.

<sup>c</sup> This enzyme is expressed as a cytochrome *oo*<sub>3</sub> complex, with little difference in activity in various overexpressing strains (Puustinen et al. 1992).

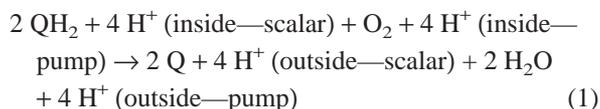
<sup>d</sup> There are two *S. acidocaldarius* enzymes; both contain cytochrome *b* homologues, yet only one contains a Cu<sub>A</sub> site (see text).

<sup>e</sup> Not determined.

<sup>f</sup> The *aa*<sub>3</sub>-type cytochrome *c* oxidase complex has been isolated from many organisms, from bacteria to mammals. See Musser and co-workers (1995) and García-Horsman and co-workers (1994a) for reviews.

<sup>g</sup> The PS3 enzyme is expressed as a cytochrome *cao*<sub>3</sub> complex under air-limited conditions (Sone and Fujiwara 1991; Sone et al. 1994).

The structural similarities of the ubiquinol and cytochrome *c* oxidase complexes naturally suggest that both families of enzymes may catalyze proton translocation via similar mechanisms. The *aa*<sub>3</sub>-type cytochrome *c* oxidase complexes have been shown to pump an average of 1 H<sup>+</sup>/e<sup>-</sup> (for a review see Musser et al. 1995). In contrast, the observed overall H<sup>+</sup>/e<sup>-</sup> stoichiometry for the cytochrome *bo*<sub>3</sub> complex has been found to be 2 (Puustinen et al. 1989). Half of the protons translocated most certainly arise from the scalar release of protons from the ubiquinol oxidation chemistry and the uptake of protons from the opposite side of the membrane to balance the dioxygen reduction chemistry electrically. It is possible, then, that a mitochondrial cytochrome *c* oxidase-type proton pump mechanism (hereafter referred to as an ‘‘ion pump mechanism’’) accounts for the remaining protons translocated by the cytochrome *bo*<sub>3</sub> complex:



Electrons are input to the mitochondrial-type cytochrome

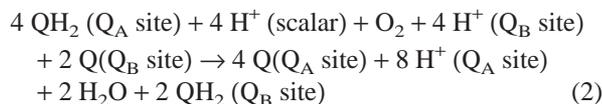
*c* oxidase complex at the Cu<sub>A</sub> site. This redox site is located in the cytoplasmic domain of the enzyme complex. Additionally, it is noteworthy that the Cu<sub>A</sub> center is the only redox site found in subunit II, and cytochrome *c* binding residues have been localized on this subunit (Iwata et al. 1995; Musser et al. 1995; Tsukihara et al. 1995, 1996). Thus, Cu<sub>A</sub> is appropriately situated to act as an intermediary acceptor of electrons from aqueous ferrocycytochrome *c* molecules, subsequently donating electrons to the hemes and Cu<sub>B</sub> in the more hydrophobic recesses of the enzyme in the membrane. In contrast, ubiquinol-8 (UQ<sub>8</sub>H<sub>2</sub>), the physiological electron donor for the cytochrome *bo*<sub>3</sub> complex, is localized in the hydrophobic interior of the membrane bilayer. Thus, the binding domain for UQ<sub>8</sub>H<sub>2</sub> is expected to be in a different three-dimensional location on the enzyme complex relative to the binding domain for ferrocycytochrome *c*. The structural differences needed to accommodate these different substrate binding sites can be easily satisfied by appropriate sequence changes. It is possible that UQ<sub>8</sub>H<sub>2</sub>, by virtue of its localization within the lipid bilayer, can approach more closely than cytochrome *c* to the hydrophobic core of subunit I containing the hemes and Cu<sub>B</sub>.

As a consequence, then, the intermediary electron acceptor,  $\text{Cu}_A$ , would be unnecessary and therefore is absent from the quinol oxidase complexes. This reasoning implies that the  $\text{Cu}_A$  site serves merely to funnel electrons into the "catalytic core" of the enzyme.

Although the dioxygen chemistry of the ubiquinol and cytochrome *c* oxidase complexes is virtually identical, the electron input mechanisms are necessarily different (Musser et al. 1993). The physiological substrate for the cytochrome *c* oxidase complexes is a water-soluble ferrocyanochrome *c* (MW ~12 kDa), a one-electron donor, whereas that for the quinol oxidase complexes is a small, membrane-associated quinol (MW <1 kDa), a two-electron donor. The electron input from ferrocyanochrome *c* to the  $\text{Cu}_A$  site occurs through a simple outer-sphere process. In contrast, the quinol oxidase complexes must stabilize the highly reactive semiquinone intermediate through strong interaction with the enzyme complex since the low-spin heme of these enzymes can accept only one electron. At present, there is little understanding of how the quinol oxidase family is able to couple the two-electron chemistry of the quinol donor to the one-electron-accepting low-spin heme of these enzymes.

Quinone(ol) [ $\text{Q}(\text{H}_2)$ ] binding proteins often have two  $\text{Q}(\text{H}_2)$  binding sites to facilitate the coupling of one-electron redox chemistry to a two-electron acceptor/donor. Typically, one of these binding sites is responsible primarily for stabilizing the semiquinone radical, whereas the other acts as the electron input/output site. Based in part on this fact, Musser and co-workers (1993) suggested that the cytochrome  $bo_3$  complex has two ubiquinone(ol) [ $\text{UQ}(\text{H}_2)$ ] binding sites. In support of this prediction, Sato-Watanabe and co-workers (1994) found a tightly bound ubiquinone-8 ( $\text{UQ}_8$ ) molecule in the as-isolated cytochrome  $bo_3$  complex that could not be removed by high concentrations of ubiquinol-1 ( $\text{UQ}_1\text{H}_2$ ) or inhibitors. These investigators concluded that the  $\text{UQ}_8$  binding site was not the ubiquinol oxidation site and that there were in fact two  $\text{UQ}(\text{H}_2)$  binding sites on the cytochrome  $bo_3$  complex. These data were interpreted to imply that the tightly bound ubiquinone acts as an intermediate electron acceptor from ubiquinol, subsequently passing electrons on to cytochrome *b*. According to this model, then, the electron input reactions of the cytochrome  $bo_3$  complex are similar to the electron output reactions of photosynthetic reaction centers. It has been found, however, that  $\text{UQ}_1$ ,  $\text{UQ}_6$ , and  $\text{UQ}_8$  as well as 2,6-dichloro-4-nitrophenol (an inhibitor) can be reconstituted into the enzyme with an empty  $\text{UQ}_8$  binding site (Puustinen et al. 1996; Sato-Watanabe et al. 1994). These data suggest that the  $\text{UQ}_8$  binding site is accessible from the solvent. In fact, Musser and co-workers (1997) have recently shown that there are two  $\text{UQ}(\text{H}_2)$  binding sites in dynamic equilibrium with the  $\text{UQ}(\text{H}_2)$  pool during enzyme turnover. It is reasonable to conclude from these data that one of the  $\text{UQ}(\text{H}_2)$  binding sites is the ubiquinol

oxidation site, whereas the other is the  $\text{UQ}_8$  binding site found by Sato-Watanabe and co-workers (1994). These data therefore support a  $\text{Q}(\text{H}_2)$ -loop proton translocation mechanism as proposed earlier (Musser et al. 1993). We emphasize that this  $\text{Q}(\text{H}_2)$ -loop mechanism fully accounts for the observed proton translocation stoichiometry (Puustinen et al. 1989) and is energetically feasible (Musser et al. 1997):

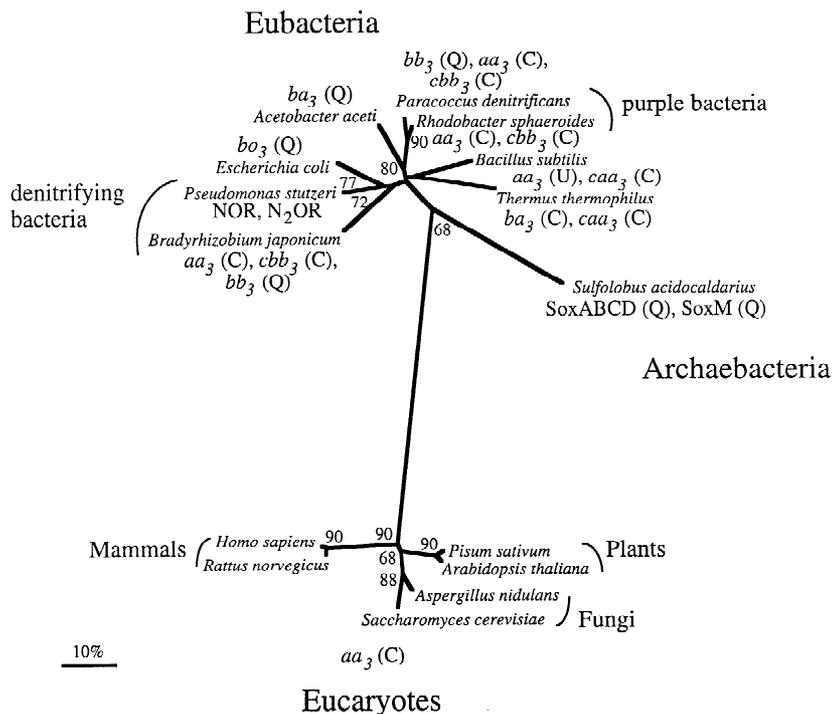


In the discussion to follow, attention is focused on the evolutionary relationship between the quinol and the cytochrome *c* terminal oxidase complexes. While it is certainly possible to argue that the  $bo_3$ -type ubiquinol oxidase complex arose from a cytochrome *c* oxidase complex that lost a  $\text{Cu}_A$  site, this viewpoint does not seem tenable from the standpoint of respiratory complexity and efficiency. Rather, the argument is made here that a primitive quinol terminal oxidase complex evolved into the more structurally and functionally complex cytochrome  $bc_1$ /cytochrome  $aa_3$  ubiquinol oxidase system.

### Phylogenetic Trees Based on Sequence Information

Phylogenetic tree construction from 16S/18S rRNA sequences is generally considered one of the most reliable methods for determining evolutionary relationships because ribosomes are universally distributed across all organisms and the 16S/18S rRNA serves functionally homologous roles in all cases. However, when such a tree is constructed, it is impossible to root the tree such that organisms containing quinol oxidase complexes and those containing cytochrome *c* oxidase complexes compose separate lineages (Fig. 1). All eukaryotes contain a mitochondrial  $aa_3$ -type cytochrome *c* oxidase complex. The eubacteria contain either quinol or cytochrome *c* oxidase complexes or both. The archaeobacteria are rooted between the eubacteria and the eukaryotes. The archaeobacterium *Sulfolobus acidocaldarius* contains two different quinol oxidase complexes and therefore severely complicates the analysis of terminal oxidase complex evolution. It is clear that alternate means must be utilized to construct the terminal oxidase complex evolutionary tree.

There are a number of evolutionary trees in the literature constructed based on amino acid sequence similarities of quinol and cytochrome *c* oxidase complexes (Castresana et al. 1994, 1995; Saraste and Castresana 1994; Schäfer et al. 1996a; van der Oost et al. 1994). These analyses are valid only if it can be safely assumed that the various terminal oxidase complexes are functionally homologous. However, the validity of this assump-

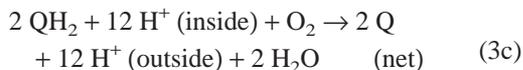
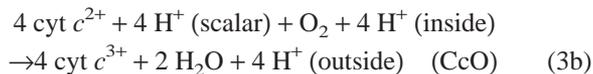
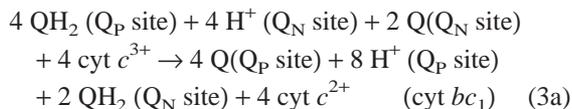


**Fig. 1.** Unrooted phylogenetic tree constructed using 16S/18S rRNA sequences from the GenBank database. The various types of quinol (Q) and cytochrome *c* (C) oxidase complexes present in the species shown are noted. Only terminal oxidase complexes that are members of the heme-copper superfamily are included. The eubacterial 16S rRNA sequences and the eukaryotic 18S rRNA sequences were first aligned as two separate groups, then aligned together. The archaeobacterial *S. acidocaldarius* 16S rRNA sequence was added to the alignment last. The tree was constructed from the distance matrix using the neighbor-joining method (Saitou and Nei 1987). Bootstrap confidence intervals (only those larger than 50% are shown) were calculated from 9,000 samples including gaps and correcting for multiple substitutions (Felsenstein 1985). All of the above calculations were made with the CLUSTAL W 1.7 software package (Thompson et al. 1994); the tree was drawn with TREEVIEW (Page 1996).

tion is called into question by the different substrates (quinol and cytochrome *c*) and the different total respiratory efficiencies of the respective respiratory chains. We explicitly take into account quaternary structural complexity and respiratory efficiency and use minimal sequence information in the analysis that follows.

## Respiratory Efficiency

The cytochrome  $bo_3$  complex combines the oxidation-reduction chemistry mediated by the cytochrome  $bc_1$  and cytochrome *c* oxidase complexes. However, a price is paid. The cytochrome  $bo_3$  complex catalyzes the net translocation of  $8\text{ H}^+/4\text{ e}^-$ , whereas the combination of the cytochrome  $bc_1$  and cytochrome *c* oxidase complexes results in the net translocation of  $12\text{ H}^+/4\text{ e}^-$ :

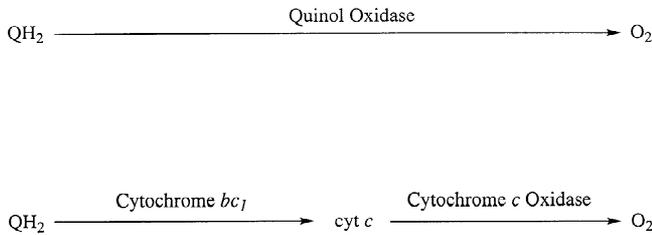


Clearly, a respiratory chain which contains the latter two enzymes is more efficient in energy transduction. However, a number of bacteria, e.g., *Paracoccus denitrificans*, *Bacillus subtilis*, and *Bradyrhizobium japonicum*,

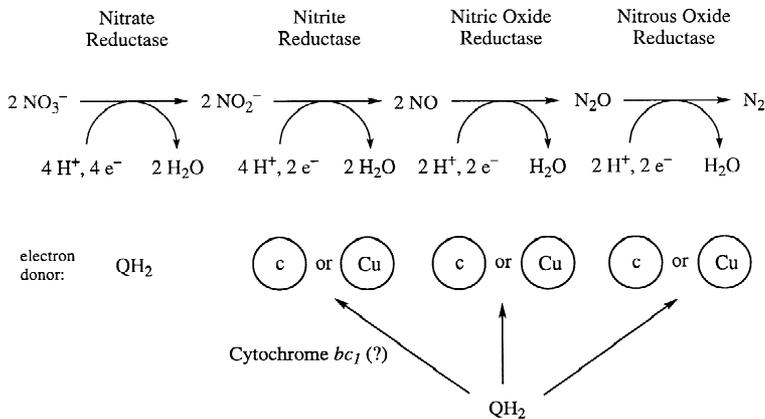
encode both ubiquinol and cytochrome *c* terminal oxidase complexes in their genome. In this manner, presumably, these bacteria are able to regulate the energy conversion efficiency of their respiratory chain at the genomic level. This regulatory capability is advantageous. When the energy supply is high, for example, ATP synthesis is unnecessary but elimination of excess reducing equivalents (e.g., regeneration of  $\text{NAD}^+$  from NADH) may become a necessity in order for other cellular functions to proceed. In contrast, regulation of the eukaryotic cytochrome *c* oxidase proton pump can be achieved through one or more of the many nuclear-encoded subunits associated with these enzymes. For example, the  $\text{H}^+/\text{e}^-$  stoichiometry of the bovine heart cytochrome *c* oxidase complex is linked to the intramitochondrial ATP/ADP ratio through a regulatory process that involves subunit VIa-H (Frank and Kadenbach 1996). This regulation or control at the protein level obviates the need for a  $bo_3$ -type ubiquinol oxidase complex.

It is typically assumed that the  $bo_3$ -type ubiquinol oxidase complex evolved from a cytochrome *c* oxidase complex that lost its  $\text{Cu}_A$  site (Castresana et al. 1994; García-Horsman et al. 1994a; van der Oost et al. 1992, 1994). However, it is difficult to rationalize why a primitive bacterium such as *E. coli* would contain this ubiquinol terminal oxidase complex, yet no cytochrome *c* oxidase complex (Trumpower and Gennis 1994), if the former enzyme evolved later. *Acetobacter acetii* contains a  $ba_3$ -type ubiquinol terminal oxidase complex very similar (67% identical) to the *E. coli*  $bo_3$ -type enzyme; like *E. coli*, *A. acetii* lacks a cytochrome *c* oxidase complex, although it does contain cytochrome *c* (Fukaya et

### Oxygenic Respiration



### Nitrogenic Respiration



**Fig. 2.** Summary of the chemistry catalyzed by oxygenic and nitrogenic respiratory chains between quinol (typically ubiquinol) and the terminal electron acceptor. See text for details.

al. 1993; Matsushita et al. 1992a, b). The respiratory efficiency of an organism that contains cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes in its respiratory chain can be reduced in two ways: (a) by uncoupling the cytochrome *c* oxidase proton pump (Frank and Kadenbach 1996) and (b) by increasing the proton permeability of the membrane with fatty acids or an uncoupling protein like that found in brown adipocytes (Rehmark et al. 1992; Ricquier et al. 1991). However, an organism which contains only a quinol terminal oxidase complex cannot *increase* its respiratory efficiency in a simple manner. In light of this argument, an organism would be at a severe selective disadvantage (e.g., during an exponential growth phase) if it had started with both a quinol terminal oxidase complex and cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes but then somehow, through evolutionary processes, lost the function of one of the latter two complexes. On the other hand, it makes evolutionary sense that the cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes arose from a primitive quinol terminal oxidase complex via a series of beneficial mutations. Unfortunately, this idea has not received the attention it deserves, largely because of the similarities between the cytochrome *c* oxidase complexes and a number of denitrifying enzymes (Saraste and Castresana 1994) and the assumption that nitrogenic respiration ex-

isted before dioxygen was plentiful enough to support oxygenic respiration (Castresana and Saraste 1995). However, one of the general rules of evolutionary tree construction is that simplicity leads to complexity. As we elaborate below, it is this basic rule that suggests that the cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes evolved from a primitive quinol oxidase complex.

Saraste and Castresana (1994; Castresana and Saraste 1995) have postulated that the *aa*<sub>3</sub>-type cytochrome *c* oxidase complexes evolved from a *bc*-type nitric oxide reductase complex. Considering the complexity of nitrate respiration, we disagree with this postulate. To illustrate this complexity, we summarize in Fig. 2 the respiratory reduction pathway of nitrate to dinitrogen by denitrifying bacteria. Of the four enzymes that make use of nitrogenous electron acceptors in this reduction process, one (the nitrate reductase complex) utilizes ubiquinol as an electron donor, whereas the other three (the nitrite reductase, nitric oxide reductase, and nitrous oxide reductase complexes) utilize a soluble cytochrome *c* or blue copper protein as the electron donor. These cytochrome *c* or blue copper electron donors are thought (in general) to obtain their electrons from a cytochrome *bc*<sub>1</sub> complex (or a functionally similar enzyme). Thus, regeneration of ubiquinone from ubiquinol requires at least five enzymes (plus the cytochrome *c* or blue copper protein electron

carrier) in nitrogenic respiration. In contrast, the reduction of dioxygen to water with ubiquinol electrons requires a single enzyme (a ubiquinol oxidase complex).

### Evolution of the Mitochondrial Cytochrome *c* Oxidase Complex

If the Q(H<sub>2</sub>)-loop model of proton translocation for the cytochrome *bo*<sub>3</sub> complex [Eq. (2)] proves to be correct, then it is apparent that not only does the cytochrome *bo*<sub>3</sub> complex carry out, in effect, the chemistry catalyzed by the cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes, but also it contains functional features of both of these enzyme complexes. This intriguing possibility intimates that during evolution, a primitive quinol oxidase complex split into two separate enzyme complexes catalyzing the same chemistry but together more efficient in energy conservation.

*Sulfolobus acidocaldarius* is a typical sulfur-dependent archaeobacterium which grows at 70–80°C and at pH 2.0–3.5 and thus appears to be a living fossil of the early biotic soup (Brock et al. 1972). This bacterium contains two similar caldariellaquinol terminal oxidase complexes of intriguing composition. One of these caldariellaquinol oxidase complexes (termed SoxABCD) contains four different polypeptides. This enzyme contains homologues of cytochrome *b* of the cytochrome *bc*<sub>1</sub> complex (SoxC) as well as of cytochrome *c* oxidase subunits I (SoxB) and II (SoxA), respectively, yet the Cu<sub>A</sub> site is lacking. The fourth polypeptide has no known homologue (Castresana et al. 1995; Lübben et al. 1992). The four hemes of this enzyme are all hemes A<sub>s</sub> (modified hemes A) (Lübben and Morand 1994). This enzyme has been postulated to translocate protons via Q(H<sub>2</sub>)-loop mechanism (Lübben et al. 1992) similar (in principle) to that described for the *E. coli* cytochrome *bo*<sub>3</sub> complex [Eq. (2)]. More recent data indicate the presence of a tightly bound semiquinone in the as-isolated enzyme and a proton translocation stoichiometry clearly in excess of 1 H<sup>+</sup>/e<sup>-</sup> (1.2 H<sup>+</sup>/e<sup>-</sup> found, on average), supporting the Q(H<sub>2</sub>)-loop hypothesis (Gleißner et al. 1997). The latter finding (>1 pumped H<sup>+</sup>/e<sup>-</sup>) indicates that the ion pump mechanism (1 H<sup>+</sup>/e<sup>-</sup>) is insufficient to explain the observed proton translocation stoichiometry (Gleißner et al. 1997). The other caldariellaquinol oxidase complex (termed SoxM) contains at least four polypeptides. One of these polypeptides is homologous to a fusion between cytochrome *c* oxidase subunit I and cytochrome *c* oxidase subunit III (SoxM), one polypeptide is a cytochrome *b* homologue (SoxG) distinct from that found in SoxABCD, and the third and fourth polypeptides contain a Rieske-type iron–sulfur center (SoxF) and a Cu<sub>A</sub> center (SoxH), respectively. This enzyme also contains four hemes: three hemes A<sub>s</sub> and one heme B (Castresana et al. 1995; Lübben et al. 1994). The possibility exists that

sulfocyanin, encoded in the same gene cluster, substitutes for cytochrome *c* as an electron donor to the SoxM complex (Castresana et al. 1995). It is not known for certain whether proton translocation occurs via a Q(H<sub>2</sub>)-loop or the ion pump mechanism (or both, since elements of both translocation mechanisms are present in this supercomplex) but it has been suggested that a Q(H<sub>2</sub>)-loop mechanism is the operational proton translocation mechanism (Schäfer et al. 1996b). The electron transfer pathways in these enzymes have not been worked out, especially with regard to the extra hemes and the iron–sulfur center. However, it is quite apparent that these caldariellaquinol terminal oxidase complexes are more complicated structurally and functionally than the cytochrome *bo*<sub>3</sub> complex and contain essential structural features of both the mitochondrial cytochrome *bc*<sub>1</sub> and the cytochrome *c* oxidase complexes.

A ubiquinol supercomplex containing both the cytochrome *bc*<sub>1</sub> and the cytochrome *c* oxidase complexes have been isolated from *P. denitrificans* and the thermophilic bacterium PS3. The *Paracoccus* enzyme consists of normal mitochondrial-type cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes as well as an extra membrane-bound cytochrome *c* (cytochrome *c*<sub>552</sub>) sandwiched between the two larger complexes (Berry and Trumpower 1985). *Paracoccus* also contains a soluble cytochrome *c* (cytochrome *c*<sub>550</sub>) that is analogous to the mitochondrial cytochrome *c*. The electron transfer path from cytochrome *c*<sub>552</sub> to the cytochrome *c* oxidase complex is apparently different from that for the soluble cytochrome *c*<sub>550</sub>, indicating an important role for both of these cytochromes *c* (Stowell et al. 1993). The ubiquinol oxidase supercomplex isolated from PS3 is very similar to the *Paracoccus* supercomplex; the major difference is the presence of a cytochrome *c* domain at the C terminus of cytochrome *c* oxidase subunit II (Sone et al. 1987). These supercomplexes are clearly highly advanced versions of a primitive quinol oxidase complex, as they contain all of the redox centers found in the mitochondrial respiratory chain. Analogous membrane-bound cytochromes *c* have been found in *B. subtilis* and *Nitrobacter winogradskyi* as well (Nomoto et al. 1993; von Wachenfeldt and Hederstedt 1990). The membrane-bound cytochromes *c* were possibly initially necessary for efficient electron transfer chemistry. Since a soluble cytochrome *c* was feasible only after bacteria acquired a second membrane, the cytochrome *c* mediating electron transfer between the cytochrome *bc*<sub>1</sub> and the cytochrome *c* oxidase complexes had to have remained membrane-bound if this membrane had not been acquired at the time these supercomplexes evolved.

A number of cytochrome *caa*<sub>3</sub> complexes have been found in bacteria. All of these complexes are similar to the PS3 cytochrome *caa*<sub>3</sub> complex mentioned above—the cytochrome *c* domain is a C-terminal extension on subunit II of the enzyme—although they have been iso-

lated without attached cytochrome  $bc_1$  polypeptides (Table 1). All of these bacteria have a soluble cytochrome  $c$  that serves to shuttle electrons from the cytochrome  $bc_1$  complex to the cytochrome  $caa_3$  complex. The functional role of the covalently attached cytochrome  $c$  is a mystery, as it appears to serve as an intermediate electron transfer site between the soluble cytochrome  $c$  and the  $Cu_A$  site, and as the mitochondrial cytochrome  $c$  oxidase complex exemplifies, this additional electron transfer site is clearly not necessary. This extra cytochrome  $c$  domain was most likely lost when mitochondria evolved and the mitochondrial genes were transferred to the nuclear genome.

### The $cbb_3$ -Type Cytochrome $c$ Oxidase Complexes

There is one class of cytochrome  $c$  oxidase complexes found in a number of bacteria (*B. japonicum*, *P. denitrificans*, *R. sphaeroides*, and *R. capsulatus*) that most have arisen within a different evolutionary branch than that which gave rise to the mitochondrial cytochrome  $c$  oxidase complex. This class of enzymes, the  $cbb_3$ -type cytochrome  $c$  oxidase complexes, contains a low-spin heme and a high-spin heme- $Cu_B$  binuclear center in subunit I of the enzyme as is common for all cytochrome  $c$  oxidase complexes. However, electron input from a soluble cytochrome  $c$  occurs through a series of three hemes in extramembranous domains of two other subunits. Notably, none of these enzymes has a  $Cu_A$  center (de Gier et al. 1994, 1996; García-Horsman et al. 1994b; Gray et al. 1994; Preisig et al. 1993; Thöny-Meyer et al. 1994). The *P. denitrificans*  $cbb_3$ -type oxidase complex translocates protons, although the mechanism appears to be different from that for the mitochondrial  $aa_3$ -type enzymes, as residues in the latter enzymes which are important for the translocation process are lacking in the *Paracoccus*  $cbb_3$ -type enzyme (see below).

### Denitrification Enzymes

The nitric oxide reductase complex from *Pseudomonas stutzeri* is clearly related to the  $cbb_3$ -type cytochrome  $c$  oxidase complexes. The largest subunit of this nitric oxide reductase complex is homologous to subunit I of the heme-copper terminal oxidase superfamily and the redox centers in this subunit are ligated by the six invariant histidines common to this family of enzymes. At first, it was thought that a heme-copper binuclear center is responsible for the reduction chemistry, but recent work suggests that a nonheme iron ion plays the role of  $Cu_B$  (Zumft and Körner 1997). In addition to the two hemes B in this subunit, this nitric oxide reductase complex contains a heme C in an associated subunit (hence, this enzyme is termed a  $bc$ -type nitric oxide reductase com-

plex) which appears to be the primary acceptor of electrons from a soluble cytochrome  $c$  (Saraste and Castresana 1994; Zumft et al. 1994). There is no evidence that this enzyme catalyzes proton translocation (van der Oost et al. 1994). *P. stutzeri* also contains a nitrous oxide reductase complex. This soluble enzyme contains a binuclear copper site that is very similar structurally to the  $Cu_A$  site of the cytochrome  $c$  oxidase complexes (Antholine et al. 1992; Farrar et al. 1991; Jin et al. 1989; Scott et al. 1989). Thus, there are at least two evolutionary links between the nitrogenic and the oxygenic respiratory pathways. As a consequence, which respiratory pathway was utilized first during evolution is certainly a provocative question.

### Proton Channels

The two X-ray structures for the  $aa_3$ -type cytochrome  $c$  oxidase complex, one for the 4-subunit *Paracoccus* enzyme (Iwata et al. 1995) and one for the 13-subunit bovine enzyme (Tsukihara et al. 1995, 1996), have allowed more explicit interpretation of mutagenesis results. One of the findings of this more informed analysis is that there are apparently at least two proton transfer paths within these  $aa_3$ -type enzymes: the K channel (so named for a conserved lysine residue) and the D channel (so named for a conserved aspartate residue) (Konstantinov et al. 1997). Interestingly, the SoxABCD complex lacks residues in the D channel; this finding is not too surprising, as the proton translocation stoichiometry ( $H^+/e^- > 1$ ) is incompatible with a pure ion pump mechanism (Gleißner et al. 1997), therefore intimating an alternate proton translocation mechanism. In contrast, the SoxM complex contains the "pump" residues (Schäfer et al. 1996b), and therefore, based on sequence comparison alone, this enzyme is predicted to translocate protons via an ion pump mechanism. The *P. denitrificans*  $cbb_3$ -type cytochrome  $c$  oxidase complex lacks presumably important residues in each channel and yet translocates protons. This finding argues for an alternate proton translocating machinery in the  $cbb_3$ -type enzymes (de Gier et al. 1996). With the present uncertainty in the actual proton transfer pathways, however, it is still too early to theorize on the evolutionary development of these pathways from the primary structure.

### Proposal for a Protein Evolutionary Tree

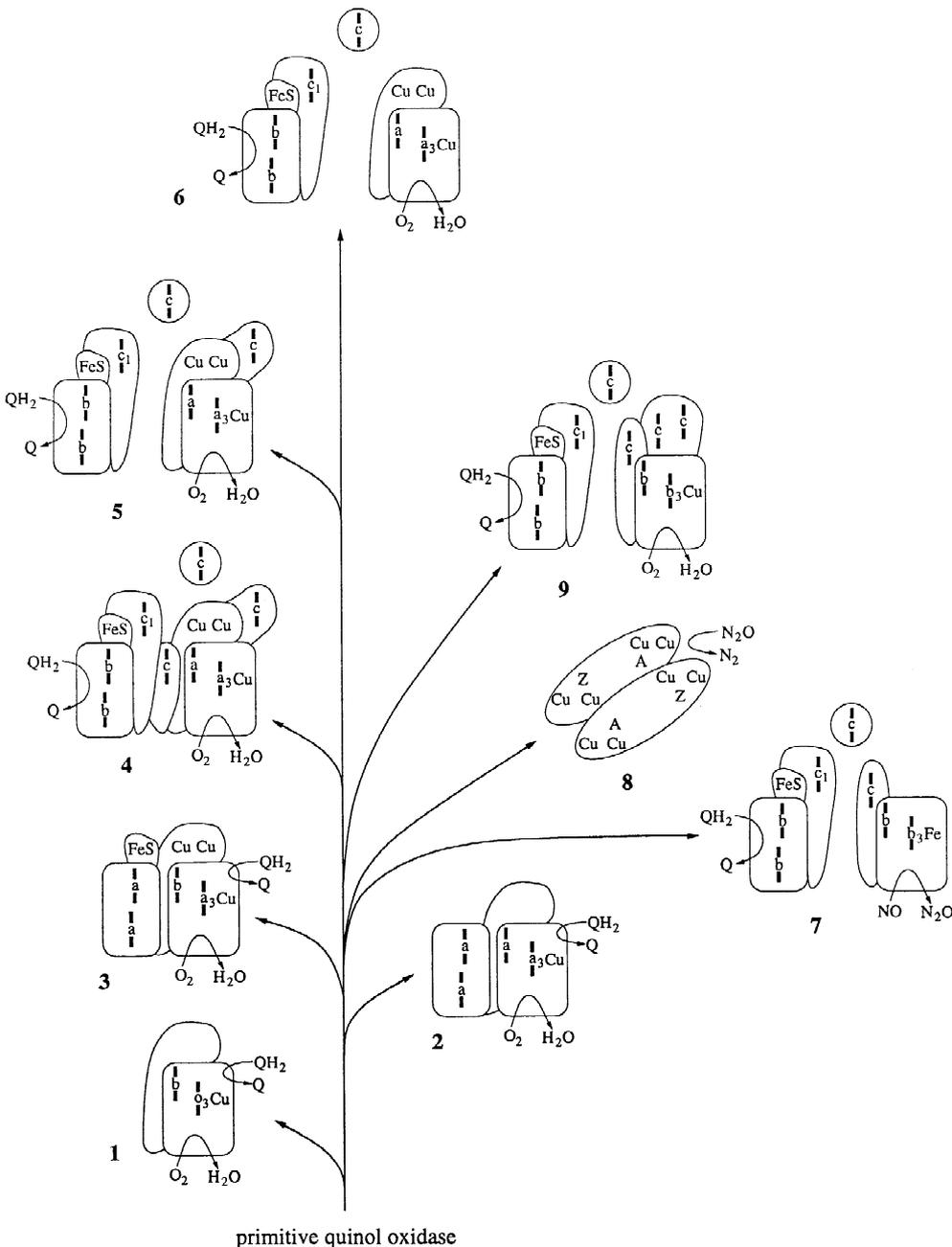
As Nature very rarely builds a new protein but rather builds with the material already at hand, it is quite unlikely that the nitric oxide and nitrous oxide reductase complexes and the cytochrome  $c$ /quinol oxidase complexes evolved independently to yield very similar structural motifs. Thus, the similar structural elements of

these nitrogenic respiratory enzymes and the cytochrome *c*/quinol oxidase complexes imply that these proteins have a common ancestor. However, it is not so simple to decipher which group of enzymes evolved first, as arguments support both possibilities. The notion that denitrifying and sulfur-reducing bacteria were present on the early earth when the availability of atmospheric dioxygen was low is an attractive one. The average temperature of the early earth was much hotter, to be sure (Bengston 1994), and the many bacteria found only in hot springs or near deep-sea thermal vents today do appear to be living fossils. However, quite a few thermophilic bacteria have quinol and/or cytochrome *c* oxidase complexes in their respiratory chains (e.g., *S. acidocaldarius*, PS3, *Thermus thermophilus*; see Table 1), suggesting that dioxygen was indeed sufficiently abundant wherever and whenever life began. In fact, it is considered possible that localized "oxygen oases" existed on the early earth despite the anoxic atmosphere as a result of water and CO<sub>2</sub> photolysis in aqueous basins with low concentrations of reduced substances (such as H<sub>2</sub>, CO, H<sub>2</sub>S, and Fe<sup>2+</sup>) (Kasting 1993; Schopf and Klein 1992). The appearance of superoxide dismutases in prokaryotes more primitive than cyanobacteria argues for an early presence of adequate supplies of dioxygen, for "it is difficult to understand the evolution of an enzyme designed to protect an organism from something that supposedly did not exist" (Bengston 1994). The many ubiquinol and cytochrome *c* oxidase complexes expressed only under conditions of low oxygen tension indicate that high concentrations of dioxygen are not required for these enzymes to function efficiently (Castresana et al. 1994). Thus, the argument that denitrifying enzymes evolved before terminal oxidase complexes because dioxygen was essentially unavailable on the early earth is not very convincing.

Natural selection works quite rapidly to eliminate mutant organisms that cannot compete or enzymes that do not provide a selective advantage. An organism that has both cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes as well as a quinol terminal oxidase complex is not expected to be able to compete successfully with other members of its species if it loses the function of one of the former complexes (e.g., due to a single-site mutation or gene deletion). The loss in respiratory efficiency will quite powerfully select against the mutant organism (e.g., during an exponential growth phase). On the other hand, a cell that is able to split its quinol oxidase activity into two complexes that yield a better respiratory efficiency can certainly father the takeover of its niche by a new species. *E. coli* do not contain cytochrome *bc*<sub>1</sub> or cytochrome *c* oxidase complexes, or anything resembling these complexes, yet this species does contain the cytochrome *bo*<sub>3</sub> ubiquinol oxidase complex (García-Horsman et al. 1994a). This discussion suggests, therefore, that a quinol terminal oxidase complex existed be-

fore a cytochrome *c* oxidase complex and that *E. coli* does not belong to the same evolutionary branch as that in which the cytochrome *c* oxidase complex arose. So far in this discussion, we have not mentioned the cytochrome *bd* complex, an alternative but structurally unrelated ubiquinol oxidase complex also found in *E. coli* (Trumpower and Gennis 1994). The most likely explanation for this second ubiquinol terminal oxidase complex in *E. coli* is convergent evolution. The cytochrome *bo*<sub>3</sub> and *bd* complexes probably evolved approximately simultaneously in two physically separated primitive cell populations. Most likely, a gene transfer event between the two populations resulted in a species with greater adaptability: the cytochrome *bd* complex has a higher oxygen affinity and is preferentially expressed under low dioxygen tension, whereas the cytochrome *bo*<sub>3</sub> complex putatively translocates twice as many protons and is preferentially expressed under high oxygen tension (Trumpower and Gennis 1994). It is interesting to note that the cytochrome *bd* complex exhibits substrate inhibition similar to that found for the cytochrome *bo*<sub>3</sub> complex (Musser et al. 1997; Sakamoto et al. 1996), therefore raising the possibility that the former enzyme also utilizes a Q(H<sub>2</sub>)-loop during turnover.

The increase in respiratory complexity throughout evolution as proposed here is summarized in Fig. 3. As every enzyme observed today is the result of approximately 4.5 billion years of evolution, it is quite incorrect to state, for example, that the *E. coli* cytochrome *bo*<sub>3</sub> complex evolved to yield the mitochondrial cytochrome *bc*<sub>1</sub> and cytochrome *c* oxidase complexes. Therefore, each of the complexes in Fig. 3 is shown at the end of a branch. However, the present-day enzymes certainly yield clues to the common ancestor of all the enzymes included in this scheme: a primitive quinol oxidase complex. Of the quinol oxidizing systems depicted in Fig. 3, the cytochrome *bo*<sub>3</sub> complex (1) is the simplest structurally and functionally and so is placed at the end of the first branch. Of the two *Sulfolobus* enzymes (2 and 3), SoxM is considered the more complex because of the Rieske protein and Cu<sub>A</sub> binding domain. The progression from the cytochrome *bc*<sub>1</sub>/(*c*)*aa*<sub>3</sub> supercomplex (4) to separate cytochrome *bc*<sub>1</sub> and *aa*<sub>3</sub> complexes (5) to the mitochondrial ubiquinol oxidizing system (6) is straightforward. There are no indications that the *bc*-type nitric oxide reductase complex (7) pumps protons, despite sufficient redox energy to support this function (van der Oost et al. 1994). As this enzyme requires the cytochrome *bc*<sub>1</sub> complex, it is placed on a branch earlier than 4 but later than 3, which lacks a fully developed cytochrome *bc*<sub>1</sub> complex. The nitrous oxide reductase complex (8), which requires the Cu<sub>A</sub> site, probably arose at about the same time as 7, as similar evolutionary pressures would select for both of these enzymes. The cytochrome *cbh*<sub>3</sub> complexes (9) catalyze proton translocation, but apparently not by the same ion pump mech-



**Fig. 3.** An evolutionary tree depicting the increase in complexity of respiratory proteins as evidenced by present-day enzyme complexes. Lengths of branches are *not* scaled to evolutionary distance. (1) The *E. coli* cytochrome *bo<sub>3</sub>* ubiquinol oxidase complex. Subunit II contains the same cupredoxin fold found in the Cu<sub>A</sub>-binding domain of cytochrome *c* oxidase and nitrous oxide reductase complexes. (2) The *S. acidocaldarius* caldariellaquinol terminal oxidase complex (SoxABCD) containing homologues of cytochrome *b* (of the cytochrome *bc<sub>1</sub>* complex) and cytochrome *c* oxidase subunits I and II. (3) The *S. acidocaldarius* caldariellaquinol terminal oxidase complex (SoxM) containing homologues of cytochrome *b* (of the cytochrome *bc<sub>1</sub>* complex) and a fusion protein of cytochrome *c* oxidase subunits I and III. The third and fourth subunits contain a Rieske-type iron-sulfur center and a Cu<sub>A</sub> center, respectively. For 1, 2, and 3, proton translocation is postulated to occur via a Q(H<sub>2</sub>)-loop mechanism (see text). (4) The cytochrome *bc<sub>1</sub>*/cytochrome *c*/cytochrome *c* oxidase supercomplex found in *P. denitrificans* and the thermophilic bacterium PS3. The C-terminal extension on subunit II of the cytochrome *c* oxidase complex (shaded) is found in the PS3 enzyme. (5) A cytochrome *caa<sub>3</sub>* complex has been found in a number of bacteria (Table 1). It is not known if a cytochrome *bc<sub>1</sub>*/cytochrome *c*/cytochrome *c* oxidase supercomplex exists under physi-

ological conditions in all cases. An active cytochrome *c* oxidase complex from *P. denitrificans* and the thermophilic bacterium PS3 can be isolated alone or as part of a supercomplex as in 4. (6) The mitochondrial cytochrome *bc<sub>1</sub>* and cytochrome *aa<sub>3</sub>* (cytochrome *c* oxidase) complexes. For 4, 5, and 6, proton translocation occurs via a Q(H<sub>2</sub>)-loop mechanism (cytochrome *bc<sub>1</sub>* complex) and the cytochrome *c* oxidase ion pump mechanism. (7) The cytochrome *bc<sub>1</sub>* and nitric oxide reductase complexes found in *P. stutzeri*. The nitric oxide reductase complex has not been found to catalyze proton translocation. (8) The periplasmic nitrous oxide reductase complex found in *P. stutzeri* contains two binuclear copper centers: a Cu<sub>A</sub>-type center, thought to function as an electron transfer site, and a Z center, thought to be the catalytic site. (9) The cytochrome *bc<sub>1</sub>* and cytochrome *cbb<sub>3</sub>* complexes found in *B. japonicum*, *P. denitrificans*, *R. sphaeroides*, and *R. capsulatus*. The cytochrome *cbb<sub>3</sub>* complexes catalyze proton translocation but likely by a mechanism different from that of the mitochondrial-type cytochrome *c* oxidase complexes. The terminal oxidase complexes at the *top* and on the *left* side of the *central line* contain both the “catalytic” and the “pump” channel residues (see text), whereas those on the *right* lack residues in at least one of these channels. See the text for references and further details.

anism as the mitochondrial  $aa_3$ -type enzymes (de Gier et al. 1996), and therefore are depicted in an earlier branch than **4** but later than the nonpumping **7**. The positioning of **7**, **8**, and **9** in this tree are clearly the most difficult. According to Musser and co-workers (1995; Musser and Chan 1995), the  $\text{Cu}_A$  center of the cytochrome  $c$  oxidase complexes is intimately involved in the proton pumping reactions of these enzymes. In contrast, recent data support a  $\text{Q}(\text{H}_2)$ -loop proton translocation mechanism for the cytochrome  $bo_3$  complex (Musser et al. 1997). Consequently, it is proposed that a primitive quinol oxidase complex translocated protons via a  $\text{Q}(\text{H}_2)$ -loop mechanism. This proton translocation mechanism has been retained by **1**, **2**, and **3**. When the evolving quinol oxidase complex split into two complexes, the  $\text{Q}(\text{H}_2)$ -loop mechanism was retained by the ubiquinol oxidation component (the cytochrome  $bc_1$  complex), whereas a new proton translocation mechanism, the  $\text{Cu}_A$ -based proton pump mechanism, developed in the dioxygen reduction component (the cytochrome  $c$  oxidase complex). This  $\text{Cu}_A$ -based proton pump mechanism developed too late to be of use to **7** and **9**. It is certainly possible for a bona fide  $\text{Cu}_A$  site to be present in **3** yet not be coupled to proton translocation. After all, the cytochrome  $b$  homologue in this enzyme complex behaves differently than its mitochondrial counterpart. Zickermann and co-workers (1997) discuss the interesting proposal that the progenitor of the  $\text{Cu}_A$  site was a blue copper site; in this mononuclear form, the copper site could have easily been required solely for electron transfer—the proton pump function arose well after the site became binuclear.

If it is accepted that the ancestral  $bo_3$ -type quinol oxidase complex was the progenitor of the cytochrome  $c$  oxidase complexes, the  $bc$ -type nitric oxide reductase complexes must have evolved after oxygenic respiration—an obvious implication of Fig. 3. This evolution scenario makes sense from the standpoint of complexity (Fig. 2). Clearly, when a cytochrome  $c$  oxidase complex existed, a cytochrome  $bc_1$  complex must also have existed. Similarly, the ferrocycytochrome  $c$  arising from cytochrome  $bc_1$  turnover is required for membrane-bound nitrous oxide reductase activity (Ferguson 1994; van der Oost et al. 1994). Once a  $bc$ -type nitric oxide reductase complex existed, the other enzymes involved in nitrate/nitrite respiration (nitrite and nitrous oxide reductase complexes) could evolve. Note that there is a nitrate reductase in *E. coli* but the nitrite produced is excreted by a nitrate/nitrite antiporter (Kaldorf et al. 1993) or converted directly to ammonia by a soluble, NADH-dependent, siroheme-containing nitrite reductase or a hexaheme nitrite reductase (Brittain et al. 1992). Attempts to find a  $bc$ -type nitric oxide reductase complex in *E. coli* have been unsuccessful (Kaldorf et al. 1993). In the context of this discussion, the absence of a nitric oxide reductase complex in *E. coli* is not surprising, as the enzyme required for generating reducing equivalents, the cytochrome  $bc_1$  complex, is also absent from this

species. Thus, according to this discussion, oxygenic respiratory complexity proceeded from a quinol terminal oxidase complex (one enzyme) to cytochrome  $bc_1$  and cytochrome  $c$  oxidase complexes (two enzymes). Along the way, nitrogenic respiration was invented involving cytochrome  $bc_1$ , nitrite reductase, nitric oxide reductase, and nitrous oxide reductase complexes (four enzymes) (note that the electron carriers between the various complexes also contribute to the increased complexity). At each evolutionary step, many pieces of the new enzymes clearly arise from structural motifs already present. In all cases, complexity and adaptability increase but respiratory efficiency never decreases. Note that this evolutionary picture does not rule out the possibility that the progenitor of the  $bo_3$ -type ubiquinol oxidase complex was a  $bo_3$ -type ubiquinol:nitric oxide oxidoreductase. Although such an enzyme has not been found in a present-day species, it could have been useful during the anoxic conditions of the early earth.

### The Terminal Oxidase Complex Tree Versus the rRNA Tree

The fact that many bacteria contain multiple-terminal oxidase complexes implies that the protein evolutionary tree shown in Fig. 3 *cannot* be converted into a phylogenetic tree. Further, as mentioned above, the 16S/18S rRNA phylogenetic tree (Fig. 1) cannot be rooted such that organisms with ubiquinol oxidase complexes and those with cytochrome  $c$  oxidase complexes compose separate lineages. Fortunately, there is an interpretation that dispels the apparent incongruence between the phylogenetic and the protein trees—namely, the transfer of genetic material between evolving bacteria. While such genetic transfer events most certainly were rare and resulted from an unusual set of conditions, it had to happen only once to alter dramatically the evolutionary direction of a particular lineage. Considering the evolutionary time scale, such an explanation is certainly plausible. Thus, we conclude that the terminal oxidase complex tree depicts the progression of a few particular enzymes' *appearance on earth*, whereas the rRNA tree depicts these enzymes' *appearance in distinct lineages*.

### Concluding Remarks

The greater functional complexity of the cytochrome  $bc_1$ /cytochrome  $c$  oxidase segment of the mitochondrial respiratory chain relative to the  $bo_3$ -type ubiquinol oxidase complex strongly argues that the common ancestor of these enzymes was a quinol oxidase complex, and an attempt has been made here to rationalize how the evolution of respiratory processes might have occurred. The basic assumption that an organism with a quinol terminal oxidase complex as well as cytochrome  $bc_1$  and cytochrome  $c$  oxidase complexes in its respiratory chain would never be able to compete successfully with other members of its species if it lost the function of one of the

latter enzymes can be experimentally tested, for example, by integrating the required genes for cytochrome *bc<sub>1</sub>*/cytochrome *c* oxidase respiration into the *E. coli* genome and determining if wild-type *E. coli* can still survive in the presence of this mutant. In an alternative, simpler experiment, one could determine which of two *P. denitrificans* deletion strains, one lacking ubiquinol terminal oxidase activity and the other lacking cytochrome *bc<sub>1</sub>*/cytochrome *c* oxidase activity [e.g., strains of de Gier and co-workers (1996)], takes over the growth medium.

More importantly, though, an evolutionary tree has been constructed on the basis of enzyme function rather than by comparing amino acid sequence information. It is considered that such an approach is required in light of the problem tackled because natural selection acts on function, not structure. If the functions of the proteins compared are sufficiently similar, an analysis of sequence alignments is expected to yield a fairly accurate estimate of the evolutionary distance between the proteins (e.g., comparison of ribosomal proteins or mitochondrial cytochromes *c*). However, if the functions of the proteins differ substantially, a comparison of primary structures for the sake of calculating evolutionary distance is fairly meaningless when little is known about the differential selective pressures on the amino acid residues of the polypeptides. The Q(H<sub>2</sub>)-loop and ion pump proton translocation mechanisms of the *bo<sub>3</sub>*-type ubiquinol oxidase and cytochrome *c* oxidase complexes, respectively, are dramatically different means whereby the same function is achieved. In addition, the fact that the *bc*-type nitric oxide reductase complex does not pump protons, whereas the structurally similar cytochrome *c* oxidase complexes do indicates that these enzymes are dramatically different functionally as well. Thus, alternative means have been utilized here to decipher the evolutionary relationship between these enzymes, namely, structural and functional complexity. From this discussion, it is apparent that oxygenic respiration is quite an old process, and, in fact, predates nitrogenic respiration as well as reaction-center photosynthesis, since both of these processes require the cytochrome *bc<sub>1</sub>* complex (or a functionally similar and homologous enzyme), which arose in the later stages of oxygenic respiratory evolution.

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